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List all unknown values. P = Rate of energy transfer Q = Energy transfer 37. Select equations. $Q = P \cdot t$ $P_{net} = \rho A_e(T_4 - T_4)$ © 2012 Project Lead The Way, Inc. Principles of Engineering Activity 1.3.3 Thermodynamics Answer Key ¶ Page 8 38. Apply known values to equations.

~~Activity 1.3.3 Thermodynamics Answer Key~~

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You do not need to show the relationship of units to solve the problem, but be sure to label your final answer as watts. Formula Substitute / Solve Final Answer(0) Pin = IV. 15*12 180W 100*12 1200W. Page 3/9. Read Book Pltw Poe Answers Activity 1.3.3 Thermodynamics Answer Key.

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Define the First Law of Thermodynamics. Thermal energy can change form and location, but it cannot be created or destroyed. List two ways thermal energy can be increased in a system. Adding thermal energy. Performing work on the system. Define the Second Law of Thermodynamics. Thermal energy flows from hot to cold. Define entropy.

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