

Inorganic Photochemistry

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Photochemistry is the branch of chemistry concerned with the chemical effects of light. Generally, this term is used to describe a chemical reaction caused by absorption of ultraviolet (wavelength from 100 to 400 nm), visible light (400–750 nm) or infrared radiation (750–2500 nm).

Photochemistry - Wikipedia

The fascinating field of inorganic photochemistry is extremely diverse. This chapter discusses some general principles governing light-induced properties of metal-containing molecular compounds.

Inorganic Photochemistry | SpringerLink

Research in inorganic photochemistry has expanded enormously in scope and importance, especially in the past decade. The development was encouraged by Balzani and Carassiti in their exhaustive monograph (I) on the photo-chemistry of coordination compounds that just preceded the decade.

Introduction to Inorganic Photochemistry

Some of the early quantitative history of inorganic photochemistry is recalled and some of the early experiences of the writer. Conceptual landmarks in the progress to the present are outlined. The role of the Italian school of photochemistry in this development is noted.

Inorganic Photochemistry - - Then and Now

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Inorganic Photochemistry

The study of the light-induced behavior of various metal compounds. The physical and chemical properties of substances are generally altered by the absorption of light. Typical metal compounds have a characteristic number (coordination number) of molecules or ions (ligands) directly bonded to the metal center.

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Inorganic photochemistry - - then and now - ScienceDirect

To learn the photoelectron spectroscopy of inorganic compounds. To study the theory, determination of structure, growth of crystals. To study the applications of IR, Raman and NMR spectroscopy in inorganic compounds To learn the detailed study of synthetic organometallic complexes and their reactivity.

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This monograph/reference focuses on those subjects that are considered essential to an understanding of inorganic photochemistry. Graduate students with a background in physical chemistry will find that the quantum mechanical treatments related to the principles of spectroscopy and chemical dynamics are readily accessible.

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