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Grade 7 Science Glossary. abiotic. physical or nonliving factor in an ecosystem. adaptation. a trait which helps an organism or species survive.

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Science 7 Textbook - Mr. Wessner's World
7th grade Life Science Glossary A Abdomen - the distinct, usually elongated posterior region of an arthropod's body Abiotic - the non-living things in an environment Adaptation - the process of becoming adapted to an environment; an anatomical, physiological, or behavioral change that improves a population's ability to survive

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Ch. 1 & 2 (1 - 57) Ch. 3 & 4 (58 - 125) Ch. 5 & 6 (126 - 183) Ch. 7 & 8 (184 - 237) Ch. 9 & 10 (238 - 311) Ch. 11 & 12 (312 - 375) Toolkit, Glossary, Index (376 - 427)

Our Gr. 7 Science Textbook
Glossary A abiotic factors [AY-bye-aw-tik FAK-turz] ? the non-living physical and chemical components of an ecosystem (p. 32) absolute magnitude ? the brightness of stars as if they were all located 33 ly from Earth; for example, Polaris's absolute magnitude is ?3.63 (p. 371)

Glossary - Nelson
Glossary How to Use This Glossary This Glossary provides the de?nitions of the key terms that are shown in boldface type in the textbook (instructional boldfaced words such as "observe," and "gather," used in investigations are not included). Other terms that are not critical to your understanding, but that you may wish to know,

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quantitative - describing how much of something there is (the cat weighs 10 kg, is 25 cm high, and has an age of 150 days) scientific method - systematic approach to problem solving. sequence - an arrangement of things or events in a particular order. theory - an explanation of things based on many observations.

7th Grade Science Vocabulary - Science Inquirer
Grade 7 science textbook glossary - How to Use This Glossary. This Glossary provides the definitions of the key terms that are shown in boldface type in the textbook (instructional boldfaced words such as .. law in science, a statement of a pattern, action, or condition Page 7 . Glossary of scientific terms for kids. of turning litmus red and reacting with a base to form a salt.

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Count atoms and molecules in chemical reactions. 3. Calculate amounts of reactants or products in chemical reactions. 4. Describe energy changes in chemical reactions. 5. Compare physical and chemical changes. 6. Explore chemical structure and properties: soapmaking.

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