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Chapter Summaries – Chemistry  
Matter and Change

Chapter 1 Section 2 MATTER and Its Properties  
Matter – Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT– a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1  
Section 1

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In your textbook, read about atomic  
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Letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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Section 3.1 Properties of Matter pages  
70–75 Problem-Solving Lab 1.

Explain why the flow of a compressed gas must be controlled for practical and safe use.

Matter—Properties and  
Changes Matter—Properties and  
Changes

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atomic mass 7. elements 8. atomic  
number 9. Henry Moseley 10. protons  
11. periodic law 12. properties 13. 14  
...

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Chemistry: Matter and Change •  
Chapter 5 25 Electrons in Atoms  
Reviewing Vocabulary Match the  
definition in Column A with the term  
in Column B. Column A Column B 1.  
The set of frequencies of the  
electromagnetic waves emitted by the  
atoms of an element 2. The minimum  
amount of energy that can be lost or  
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## Chapter Assessment

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Change. extensive property. intensive  
property. extensive properties  
(examples) intensive property  
(examples) a property that depends  
on the amount of matter in a sample.  
a property that depends on the type  
of matter in a sample not.... mass,  
volume, length, weight.

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Chapter 11 31 number of moles of  
known substance C H C H C H O O O

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CO CO CO H O H O H O number of  
moles of unknown substance O CO H  
O C H CO H O C H O H O C H O CO  
use mole ratio 5 mol O 2 1 mol C 3 H  
8 3 mol CO 2 1 mol C 3 H 8 4 mol H  
2 O 1 mol C 3 H 8 1 mol C 3 H 8 5  
mol O 2

## MATH SKILLS TRANSPARENCY MASTER 16

A “ chemical change ” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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