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Analysis The mass of the water in the tank and the total mass are $m_{\text{tank}} = 3 \text{ kg}$, $V = 0.2 \text{ m}^3 = 200 \text{ kg}$, $m_{\text{total}} = m_w + m_{\text{tank}} = 200 + 3 = 203 \text{ kg}$. Thus, $1 \text{ N} = 1991 \text{ NW} = mg = (203 \text{ kg})(9.81 \text{ m/s}^2) = 21 \text{ kN}$. H₂O 1-21-8 The interior dimensions of a room are given.

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(PDF) Thermodynamics-an-Engineers-Aproach-Yunus-Cengel-6th-edition-SOLUTION-MANUEL | Kaya Ba ar Tiyekli - Academia.edu Thermodynamics 1-1C Classical thermodynamics is based on experimental observations whereas statistical thermodynamics is based on the average behavior of large groups of particles.

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